

Chemistry 214, Quantitative Analysis Lab Summer 2023 Syllabus

Chem 214-001 Quantitative Analysis Lab (1 credit hour) Summer Session A (6-weeks): May 22nd – June 30th, 2023

Tuesdays and Thursdays 8:30 am - 12:30pm

<u>Prerequisite:</u> Chem 106/102 and 112, as well as active attendance or completion of lecture Chem 212.

Lab Location: Flanner Hall 313

Laboratory Coordinator: Dr. Katrina Binaku

Office: Flanner Hall 304 | Email: kbinaku@luc.edu | Phone: (773) 508-8715

Office Hours in person (@ my office): Tuesdays & Thursdays 12:30 – 1:00pm and by a scheduled appointment.

Office Hours in ZOOM: Wednesdays 9:00 – 9:30am and 12:00 – 12:30pm.

<u>Teaching Assistant (TA):</u> Bailey Hanson. See Sakai → Syllabus tab for contact info and office hours.

<u>Course Meeting Times:</u> This course is a 100% in-person laboratory. There are no online substitutions/work for in-person laboratory work. No course modifications will be made. <u>Attendance is mandatory</u>. This course has synchronous (real-time, in-person lab experiments) work and asynchronous (online, out of lab work including recorded lectures, independent videos and/or virtual online activities, quizzes, etc.). There are no excused absences; do <u>not</u> plan work or vacations or other things during scheduled class times. There are limited to no opportunities to make up lab experiments if missed due to absences; that is because of the accelerated nature of this summer course. If absence is due to a claim of contracting COVID-19 or another illness, the lab coordinator will ask to see proof of a positive test and/or doctor's visit. In fact, any absence will require document PROOF of the reason. IF any makeup time is offered for missing a lab experiment, it is done on the lab coordinator's terms and only during scheduled class time.

Summer courses are optional; registering for a summer course means students are aware of the high expectations and lack of course flexibility. This course fits an entire semester of laboratory in to 6-weeks. It is the student's responsibility to pay attention to all information regarding the course, including the course schedule which is at the end of this syllabus. As a student enrolled in the course, you agree to be 100% committed to follow and complete all course aspects including requirements, experiments, lab report & peer review, assignments, homework, quizzes/exams, abide by due dates, rules, etc. set forth in this syllabus and displayed in Sakai. This course requires your full commitment so make sure you can commit 4-5 hours per week to complete the course homework activities outside of the scheduled lab time. All times listed are Central Standard Time (CST); all due dates are in CST. Expect the experiments to take the full 4-hours to complete!

<u>Academic Calendar:</u> It is the student's responsibility to not only know the schedule for this course but also the official <u>University Academic Calendar</u> and important dates in the calendar.

Welcome to Chem 214. I look forward to having you in the course! Check Loyola email & log-in to Sakai often.

Course Description:

This lab course emphasizes application of some topics covered in the lecture course (Chem 212). The course emphasizes laboratory and chemical safety, introduces students to classical and modern methods of chemical analysis, wet chemical laboratory techniques, demonstrates use of Excel for basic statistics and experimental data analysis, and exposes students to real-world samples to be prepared for future use of lab techniques and instrumentation. Topics covered will include Microsoft Excel, basic statistics and data analysis, acid-base titration, EDTA titration, buffers, pH titration curves and derivative graphs, a module on chromatography focused on High-Performance Liquid Chromatography (HPLC), lon Chromatography (IC), and Gas Chromatography (GC), and a UV-Vis spectrophotometry module with a focus on external standards and standard addition calibration methods to answer chemical questions about two analytes. Students will also be introduced on how to evaluate an analyte using ATR-FTIR spectroscopy. Chemical knowledge spanning from first-year chemistry, organic, to new topics in Chem 212 lecture is vital. This list is not exhaustive but mentions the highlights. To be successful in any course [including this one], an honest effort and time commitment on the students' part is vital. Students must choose to commit to learning, adequately manage time, complete course work, and ask for assistance from lab coordinator and TA when struggles arise.

Course Goals & Outcomes for Students:

Goals:

- 1) Demonstrate proficiency in Microsoft Excel capabilities for data organization, graphing, data analysis, and descriptive statistics; ID the importance of accuracy & precision of laboratory work
- 2) Acquaint students with common classical and modern techniques in analytical chemistry
- 3) Show students classical conventional data collection and instrumental data similar to what is gathered in both commercial and academic laboratories
- 4) Convey importance of interpretation and evaluation of experimental results, as well as being able to effectively report experimental results through scientific writing

Outcomes:

- Use Microsoft Excel to organize and analyze data via descriptive statistics; generate graphs and figures up to the standards of scientific publications
- Evaluate accuracy, precision, and validity of experimental data
- Demonstrate proficiency in the set-up of lab equipment and completion of experiments using classical and modern techniques; explain how changing instrument conditions [or experimenter errors] affect data
- Articulate experimental results by keeping a notebook and in the format of scientific writing through lab reports

Teaching Assistant (TA) Role:

TAs help the Lab Coordinator facilitate learning, deliver course content, and provide help to students. TA monitors the laboratory during in-person labs, ensuring a safe and productive environment. TA will help students develop critical thinking and problem-solving skills. TA responsibilities also include holding an office hour, grading some course materials, and answering student questions via email. Dr. Binaku and TA are in constant communication and "CC" each other on email replies to students. This mitigates a student emailing both Dr. Binaku and TA with the same question; one reply is given and will be the same answer whether from Dr. Binaku or TA. Utilize both Dr. Binaku and TA for assistance. Note, Dr. Binaku has final authority in all matters relating to the course.

Email Etiquette:

When sending emails to Dr. Binaku or TA, please put Chem 214, section # in the email subject line or there will be a delay in response. Dr. Binaku teaches multiple courses and must know which course a student is in before replying to email. Weekday emails will get a response within a few hours; I do teach a lab right after ours from 1:30 – 4:15pm, so do not expect an email response during that time. Emails after 7:00 pm may not be replied to until the next morning. Dr. Binaku checks email on weekends; response times are longer [< 24-hours]. TA may only check email once on the weekends; be advised.

Required Materials:

1. Dressing appropriately for laboratory work, use of chemicals and glassware. See Footwear/Clothing section.

- 2. Long-sleeve lab coat [white preferred, but any color ok]. You must purchase this [LUC Bookstore or Amazon].
- 3. Lab goggles. Lab Coordinator will provide 1 free pair of goggles on day 1!
- **4.** Composition style notebook (not spiral bound & no tear-out perforations). Line ruled. You must purchase this.
- 5. Chem 214 Lab Manual. Provided for free as a PDF in Sakai. Lab Coordinator will print 1 copy per student for free.
- **6.** CamScanner app, for iPhone or Android. This is a free app that will convert a phone picture to a PDF file. It is necessary that you take pictures of your Composition notebook pages and upload them for grading as a PDF file. You may also use the Notes app (iPhone) OR a scanner machine, there are several in the Information Commons.
- 7. A non-erasable pen. Pencil and white out are not allowed.
- **8.** Scientific OR graphing calculator. Suggested model: CALC TI30XA SCIENTIF/STAT FRAC. A graphing calculator is o.k. too. Cell phones are not a substitute for calculators; do not use them.
- **9.** <u>Sakai access</u> (free for LUC students) via the internet to review and complete course content, access resources, review grades, etc. Make sure your internet connection is stable.
- **10.** Desktop or Laptop computer. Instrument simulation webpages may not work on tablets nor mobile devices and Sakai does not display well on them. If you do not have a desktop or laptop computer, there are plenty in the Information Commons. Also see extended loan equipment program if applicable. Lab Coordinator is not responsible for coordinating this resource for students nor responsible for loaned device.
- **11.** Panopto (free for LUC students). One format of recorded course content is Panopto videos. You may be prompted to log in with UVID username and password to view the videos. Links to videos will be provided in Sakai and via email when necessary.

Mask Requirement:

Masks are optional. Consider the absence policy and the fact there are *very limited* opportunities to make up an inperson lab experiment. Masks mitigate illness/spread of it. It is course policy that if during the term the University reinstitutes/reverts back to required mask wearing, the class will do so immediately.

Footwear/Clothing:

Closed toe, closed heel shoes required [no sandals, flip flops, slippers, Crocs, ballet flats, boat shoes, perforated shoes, etc.] No skin on legs, ankles, or feet can be exposed. Long pants recommended. Shorts, skorts, kilts, and skirts [unless floor length] are not allowed. Any shirt type is ok because a lab coat is worn. Bare skin on the lower extremities is a safety hazard: Be advised, concentrated acids/bases will be used in some lab experiments. Lab coats, goggles, mask, and gloves are required and must be worn at all times. This even applies when cleaning glassware! Lab coats must be fully buttoned to be an effective shield against chemicals. Students will be sent home if proper clothing or footwear is not worn, this counts as an absence. A safety lecture will be given the 1st day of class; this lecture is required to perform lab experiments. Students will sign a lab safety sheet acknowledging their understanding and commitment to adherence of lab safety rules/policies. If a student is absent the 1st day and misses the safety lecture, they cannot perform wet chemistry until the safety lecture is completed & safety sheet is signed. It is advised students do not wear contact lenses in the laboratory, as contact lens material may react with chemicals/ chemical vapors if they get into the eye. All rules are meant to keep students safe in the laboratory. Lab Coordinator and TA have complete discretion to prohibit a student from completing lab work if the student has clothing/footwear exhibiting a potential safety hazard OR exhibits behavior deemed unsafe to self or students OR exhibits any indication of being under the influence of any substances.

General Policies:

- Acting in a safe manner, wearing proper clothing and PPE, respecting chemicals, glassware, and equipment is
 required. Watching assigned Panopto videos BEFORE coming to lab is essential for knowledge of the
 experiments. Knowledge = safety. If Panopto experiment videos are not watched at least to 90% completion [do
 not fast forward more than 1.5x speed as I will know], student will not be allowed to complete an experiment
 and that is a 0. Unpreparedness can lead to endangering oneself or others in the laboratory.
- Attendance is mandatory. Points can only be earned when present in the laboratory to complete the experiments and analyze one's own data. "Life happens" so if you must be absent contact Dr. Binaku right away.

Due dates for course work [quizzes for example] are not adjusted for absences. All assigned items have specific due dates. Showing proof of absence reasons may be required.

- Most of the lab experiments must be completed as an individual. This course design ensures students learn the skills first-hand in the laboratory as an individual, essential for real-world work. Speaking from experience, when working in industry and in graduate school Dr. Binaku analyzed 100s of samples per week and could only rely on her own lab skills learned. She had no lab mates nor research partners. It is very important that students develop their own individual laboratory skills. There may be an option to work with a partner for one lab experiment, since collaborative working environments are valuable too.
- Other than office hours or a scheduled appointment, the synchronous sessions are the only other "real-time" opportunity to ask questions and communicate. Emails work great, but they are not "real-time." There is a delay with an email reply. Keep that in mind.
- Be mindful that everything in the course has a due date. Generally, graded course work cannot be made up if missed. See late work policy in the Grading section of the syllabus.
- Course work will be graded with an emphasis on correct significant digits, consistent results (do data & observations match conclusions), correctness of calculations, data analysis, or optimal instrument conditions for analyte applications, appropriate use of Excel functions, and thoroughness in responses. Following directions of reporting calculated answers are taken into account too.
- The Composition notebook needs to contain all laboratory experiment information [Date, Title, Intro, data/observations/calculations, and conclusion for an experiment]. Use the notebook as a resource. Feel free to take class notes in it to organize your thoughts too. The Lab Coordinator and TAs see the lab notebook pages when uploaded to Sakai as a PDF using the Cam Scanner app mentioned or scanner machine.
- Aspects of course work must be completed in the avenue/medium that they are provided in and in the time
 allotted [i.e. be mindful of due dates]. This means that a quiz in Test and Quizzes in Sakai can only be submitted
 in Sakai Tests & Quizzes or a notebook entry can only be submitted in Sakai Assignments. Course work items
 such as homework, quizzes, lab experiment results, lab simulation results, lab report, etc. can never be
 submitted via email. No exceptions. Submit work in required, respective medium, on time. This allows Lab
 Coordinator and TAs to see the submitted work in a central location in Sakai and therefore grade items quickly.
- Be mindful that everything in the course has a due date. Generally, graded course work cannot be made up if missed. See late work policy in the Grading section of the syllabus.
- There is a point value associated with the work, and one cannot earn points for work not completed. Students are expected to complete all course work; no makeup work is given. Not completing work for 2 or more of the in-person lab experiments or online activities is significant and unacceptable and will result in academic failure.
- Students should not enroll in courses that they cannot fully attend. Missing 2 or more lab experiments will result in academic failure regardless of the course grade at the time of the infractions.
- Dr. Binaku will ask how things are going over. I care a lot about students, course content, and your progress! Also, TA presence should enhance the educational experience. If not the case, talk to Dr. Binaku at any time.

Recording Policy and Course Content Policy:

- Panopto is used to record pre-lab lectures and content information for the course. This content is posted to the Sakai site. Lab Coordinator does not intend to record any of the in-person lecture notes while in laboratory. If any content is recorded by Lab Coordinator it is done outside of class time and no students are included.
- The use of all video recordings will be in keeping with the University Privacy Statement shown below: Assuring privacy among faculty and students engaged in online and face-to-face instructional activities helps promote open and robust conversations and mitigates concerns that comments made within the context of the class will be shared beyond the classroom. As such, recordings of instructional activities occurring in online or face-to-face classes may be used solely for internal class purposes by the faculty member and students registered for the course, and only during the period in which the course is offered. Recordings are not shared outside of this course. The above bullet point states when recordings will occur in this course (synchronous

- sessions). Recordings including student activity that have been initiated by the Lab Coordinator may be retained by the Lab Coordinator only for individual use.
- All activities pertaining to the course should be completed as an INDIVIDUAL. Any collaboration on course
 material and/or graded materials can constitute cheating. Failure of the course may result if an instance of
 copying or sharing answers to graded content is discovered by TA or Lab Coordinator.
- Chegg, Course Hero, Reddit, among other webpages, are monitored by the Lab Coordinator. If any Chem 214 course content is posted on these sites or other, the Dean and University will be notified. Student(s) involved may fail the content the posted material pertains too and/or fail the course. Posting any course content online to facilitate getting answers is a form of cheating and will not be tolerated. These websites readily give up student information to Universities as evidence of cheating/posting content that does not belong to the student.

Smart Evals:

Feedback on the course is important so that a Lab Coordinator can gain insight into how to improve the course, the teaching style, and so the department can learn how best to shape the curriculum for future semesters. Towards the end of the term, students will receive an email from the Office of Institutional Effectiveness with a reminder to provide feedback on the Chem 214 course the student is enrolled in. This office will send you reminders during the open period of feedback until the evaluation has been completed. I do read the Smart Evals and thank you in advance for completing it! The evaluation is completely anonymous. When the results are released, no one will be able to tell which student provided the feedback. Feedback is not released until after the term is over; feedback given will not impact grades.

Blanket Statement About "technical difficulties" with Technology and/or Software:

It is strongly encouraged that all required submissions to Sakai as well as typing lab reports, opening course/data/experiment files, be completed on a reliable wired or wireless internet connection. WI-FI is perfectly o.k. if the connection is reliable. The internet user must determine the reliability of their WI-FI. Excuses of "technical difficulties" are generally not accepted as this syllabus is stating all students should ensure their internet connection is reliable [not prone to outages]. The Lab Coordinator realizes campus is closed and University computer labs may not be accessible. Even so, students should ensure their internet connection is reliable enough to complete an online course without interruption. If an outage arises, the Lab Coordinator does reserve the right to ask for proof. The best advice the Lab Coordinator can give is to NOT complete assignments at the last minute, so to avoid glitches with internet, since every part of the course work needs reliable internet to submit. Lab Coordinator is not responsible for technical difficulties of personal devices [phone, tablet, home/work/public wireless internet or computer]. Do not submit items in Sakai using a cell phone or a tablet device as these devices do not count as reliable internet connection tools [and the Sakai website display on these mobile devices isn't reliable]. This syllabus is stating all students should use a reliable internet to submit work in Sakai, take quizzes, type & submit lab reports, open course/data/ experiment files. Emailing lab reports, notebooks, lab results, or other is not allowed in place of the required means of turning in lab reports or required submission of items in Sakai. This list is not exhaustive and do note that any activities this course may require a computer or internet connection for should be completed using University computers with wired internet connection.

Loyola University Absence Policy for Students in Co-Curricular Activities (including ROTC):

Students participating in co-curricular activities must make information concerning time conflicts with University sponsored events available to the Laboratory Coordinator no later than 1 calendar day after the start of the term. The Laboratory Coordinator reserves the right to contact the Athletics Department confirming time conflicts, absence, and regarding concerns. Students missing classes while representing Loyola University Chicago in an official capacity (e.g. intercollegiate athletics, debate team, model government organization) will need to discuss their needs with the Laboratory Coordinator. Sakai work cannot be made up in any circumstances, no exceptions. Laboratory work generally cannot be made up, but this will be decided case by case when appropriate. If Lab Coordinator has PowerPoints for the day/content missed, they are posted to Sakai for student access 24/7. These types of absences are handled on a case-by-case basis with remedy. Students should discuss with faculty the potential consequences of missing class and the ways in which they can be remedied. Students must provide Lab Coordinator with proper documentation describing the reason for and date of the absence. This documentation must be signed by an appropriate Faculty/Staff member, and it must be

provided as far in advance of the absence as possible. It is the responsibility of the student to make up any assignments under the timeline the Lab Coordinator decides upon.

<u>Loyola University Absence/Accommodations Policy for Religious Reasons:</u>

Students missing an in-person lab experiment due to observing religious holidays must alert the Lab Coordinator no later than 1 calendar day after the start of the term to request a special accommodation. This is handled on a case by case basis. The Lab Coordinator reserves the right to contact Campus Ministry, which keeps information on a plethora of religions. Students must plan ahead for online lab activities; work ahead if a due date is on a religious holiday. Students must discuss with the Lab Coordinator the consequences of missing lab and the ways [if any] they can be remedied, while also providing the Lab Coordinator with proper documentation describing the reason and date of the absence. The document must be signed by an appropriate Faculty/Staff member, and it must be provided as far in advance of the absence as possible. It is a student's responsibility to proactively ask what will be missed due to absence.

Course Repeat Rule:

Effective with the Fall 2017 semester, students are allowed only THREE attempts to pass Chemistry courses with a C- or better grade. The three attempts include withdrawals (W). After the second attempt, the student must secure approval for a third attempt. Students must come to the Chemistry Department, fill out a permission to register form or print it from the Department of Chemistry & Biochemistry website: http://www.luc.edu/chemistry/forms/ and obtain a signature from the Undergraduate Program Director, Assistant Chairperson, or Chairperson in Chemistry. A copy of this form is then taken to your Academic Advisor in Sullivan to secure final permission for the attempt.

Previous Course Work When Repeating the Course:

If you were enrolled in a Chem 214 lab course in a previous semester are re-taking it for any reason [withdraw, drop, unfavorable grade, etc.], please note that any coursework, data, etc. from a previous semester does NOT count in the current semester the course is being taken. Students must complete all coursework within the same semester of the enrolled course. Any previous data or coursework taken in prior semesters does not count in the current semester.

Pass/Fail Conversion Deadlines and Audit Policy:

A student may request to convert a course into or out of the "Pass/No-Pass" or "Audit" status. For Summer I 2023, students are able to convert a class to "Pass/No-Pass" or "Audit" through Friday, May 26th. Students must submit a request for Pass/No-Pass or Audit to their Academic Advisor.

Accommodations via Student Accessibility Center (SAC) Policy:

If you have a documented disability and wish to discuss academic accommodations, <u>discuss with the Lab Coordinator</u> as soon as possible, ideally the first week of class. The Coordinator of Student Accessibility Center (SAC), formerly referred to as SSWD, is located in the Sullivan Center and must be contacted independently by you, the student. Necessary accommodations will be made for students with disabilities who procure a SAC letter. However, to receive any accommodations self-disclosure, proper documentation, and registration with the SAC office at Loyola University Chicago is required. Accommodations cannot be made until the Lab Coordinator receives proper SAC documentation. Furthermore, accommodations are not retro-active and begin only once appropriate SAC documentation has been received by the Lab Coordinator in a timely manner. Only those accommodations that are specifically listed in the formal SAC letter will be provided. If a SAC letter suggests the Testing Center be utilized, it does not apply for this course as there are no written exams. Read up on <u>SAC Policies and Procedures</u>. Understand that accommodations may not be applicable during a summer course due to the condensed format. Extended/excused attendance cannot be honored in a summer course, nor can extended deadlines/due dates.

Laboratory Procedures:

PowerPoints will serve as the basis for theory, application, instrument information, and instruction for in-person lab experiments as well as online activities. A lab manual is provided, containing information about each of the in-person lab experiments and online lab activities. Students are expected to read lab procedures *several times* before coming to lab, to comprehend and complete labs safely in the laboratory and watch any provided Panopto video information about the

topics. The pre-lab lecture slides are provided in Sakai, for further information. Other media for demonstration and topics includes JoVE videos, original experiment videos, virtual simulations, etc. Other handouts will be provided as needed. All information and resources are posted in Sakai.

Lab Experiments/Activities:

All in-person lab experiments and online simulated lab experiments, activities, and/or data analysis are completed individually by students, emphasizing development of an individual's skillset. The experiment/activity topics are located in the lab schedule at the end of this syllabus. Each laboratory topic is approached uniquely. For the in-person lab experiments, students must be present in the laboratory to complete the experiments and collect the necessary data to satisfy the purpose of the experiment. Students will have their own "real-word" or "unknown" sample and have to process [experiment with] it, collect and analyze data, and report final results. There is no substitute for the in-person lab work. For some online lab activities there may be required protocol videos to watch. For other online activities that utilize a virtual simulator of the technique or instrument, students may have to develop a protocol, define instrument parameters for the best result of the instrument function, change instrument parameters and discuss the results, to name a few. All of this mimics instrument technique, method development, and/or other key skills needed in the realworld. This combined with the in-person lab skills learned will set students up with a fantastic foundation of laboratory and data analysis skills/knowledge. Each lab experiment/activity is completed in the order shown in the lab schedule. If the lab experiment/activity requires data analysis [calculations, etc.] students report the work VIA Sakai Assignments, their data of each individual determination (trials), mean/average concentration (or percent composition), standard deviation, and parts per thousand (ppt) associated with the overall determination. Students are NOT permitted to repeat/redo a lab experiment/activity. No exceptions.

If the Lab Coordinator finds a calculation error in submitted work, has to ask a student to double check their work due to invalid results, or finds an uploaded results file cannot be opened in the student's Sakai submission a 5 to 10 point **deduction** is applied to the grade. A student <u>must</u> submit revised work if Lab Coordinator finds a mistake/errors in the calculations/results. If a student finds a mistake in their own work and has to request to Lab Coordinator that an additional submission be granted, a 5 to 10 **deduction** is applied to the accuracy grade. Therefore, ask Lab Coordinator and TA questions before submitting results in Sakai. We offer lots of assistance when we know you need the help. All Sakai Assignments lab experiment/activity results submissions have directions for what to submit.

Laboratory Notebook:

One notebook is required. A bound Composition style is needed as pages are bound [can't be torn out]. Complete notebook pages in PEN. Detailed notebook requirements are in a document in Sakai Resources; this process requires electronic submission of notebook pages as a PDF file for grading purposes. That way students physically keep their notebooks 24/7 and will use Cam Scanner or scanner machine to digitally upload their notebook pages for grading.

All in-person lab experiments and online lab activities must have a complete notebook entry written in the lab notebook. Several notebook entries will be formally graded. Students will use the app called Cam Scanner OR a scanner machine to take pictures of the notebook pages for submission to Sakai as a PDF file. If the lab data has Excel components you do NOT have to put Excel in the physical lab notebook, but you need to show some example calculations written in the lab notebook [even if the calculation work was done via Excel]. Make a note.

Laboratory Report:

Lab reports must be computer generated [typed] and follow the format defined the documents in Sakai Resources. Formal lab reports are to be completed individually. Plagiarizing other students' reports (current or former), book or

internet sources, or lab procedures will not be tolerated. <u>You CANNOT copy course resources word for word; that is plagiarism</u>. Cite course resources and outside sources when applicable in a Reference section in the report.

Lab report due dates are located in the laboratory schedule. Lab reports are not accepted via email. Reports must be submitted to the appropriate Sakai Assignment as a Word Doc or PDF file within the first 15 minutes of the official lab start time on the date the report is due. Reports will be checked for plagiarism via Turn It In software. One cannot show TA or Lab Coordinator a lab report on a laptop or other device; that does NOT count as turning in a lab report on time. If a student is absent on the day a lab report is due, the report must still be turned in. If a student turns in the incorrect lab report i.e. a lab report that is not the required report(s) listed no credit is given so a zero (0) is recorded and the student is offered an opportunity to turn in the correct report, but it is considered late based on the late lab report policy. Late lab reports will receive a 4pt penalty deduction each day the report is late and result in a grade of zero (0) if not received within one week of the due date. "Day" is defined as the 7 days of the week Monday to Sunday. Late penalty applies to all students, including absent students. This means the weekend days count as late days i.e. if a student has an item due Friday but doesn't turn it in until Monday that is technically 3 days late. If a student has an item due Wednesday but doesn't turn it in until Monday that is 5 days late.

Writing skills are important to explain results and other important information in the "real world," but the Lab Coordinator realizes completing lab reports is labor intensive. Students will only write a lab report for one (1) of the lab experiments/activities in this course.

The following list includes the lab experiments for which a written lab report is required:

1) Acid-Base Titration: KHP and NaOH. Determination of % KHP in an Unknown

To assist students in improving writing skills and address any deficiencies, the Acid-Base Titration lab report, students will complete a peer review on a rough draft of another [anonymous] student's formal lab report. Then a student will be able to see the critique/review of their own report, make revisions, and submit a 'final' draft for grading by the TA or Lab Coordinator. This experience is on par with what scientists who publish work in international journals experience.

Laboratory Quizzes (Tests & Quizzes):

There will be a short quiz on the content for some lab experiment/activities. A quiz can only be taken once. Quizzes are open for a week, then close and cannot be re-opened. Take it as an individual (no help from others as that is cheating). You ARE allowed to use any resources you want. Quizzes may have questions on lab experiment/activity background information, calculations, error analysis, to name a few. Grades for each quiz are released after all sections take the quiz and it closes. If you complete a quiz early, you won't see a grade/feedback until all classmates complete it. Absent/ill students do not get extensions on quiz deadlines. Due dates are posted in Sakai and the lab schedule at the end of this syllabus.

Laboratory Safety Points:

Unsafe actions in the lab are NOT tolerated. All students start with 20 safety points. Deductions are taken for being late or unsafe actions in the lab. A student is told when a safety infraction is witnessed by TA/Instructor and that safety points were deducted. This is documented on the sign-in sheet. **Potential safety point deductions:*** Coming late to lab, not signing the sign-in sheet when present, not wearing or needing to borrow borrowing lab goggles or a lab coat, eating/ drinking in lab, chewing gum, taking goggles off in FH-313 when chemicals/glassware are still on any of the 3 lab benches (even if not your chemicals/lab bench), taking mask off or putting mask below nose at any point in the lab, not wearing goggles during lab, touching face/cell phone/personal belongings with gloves on, leaving lab with gloves on, not cleaning up chemical spills on bench top/balances/fume hood, standing/kneeling on chairs, improper chemical disposal,

not starting clean-up on time, etc. *The list is not exhaustive; if an [unlisted] action is unsafe, a student will lose safety points.

Lab Clean-up:

Students are REQUIRED to begin cleaning <u>10 minutes</u> before the official end time of class listed in LOCUS. Students are not allowed to stay past lab time to do wet chemistry under any circumstances NOR can a student gain access to the laboratory room, FH-313, outside of the LOCUS scheduled class day & time.

Academic Integrity:

The standard of academic integrity and personal honesty delineated in the College of Arts & Sciences Statement on Academic Integrity. Integrity is expected of every student and will be enforced. Cheating can take many forms in a lab course, but the most common forms are copying data/data analysis, answers to analysis questions, sharing files, or completing Sakai work or other electronic content with another person. The data and analysis, homework, quiz answers, etc. submitted for grading must be your own. If it is not, no credit will be awarded, and no make-up work for those points will be granted. Findings of dishonest academic behavior are reported to the Chair of the Chemistry Department and to the Dean's Office; it is also entered into an individual's record. Copied answers to course work or copied formal lab reports will result in penalty for all students involved. Turn It In is utilized for formal lab reports to identify plagiarism, cheating, and other. Students can converse, brainstorm, and work through strategies together but copying other students' (current or previously in Chem 214) work and presenting it as one's own is unacceptable. There is a difference between sharing knowledge and cheating. If lab reports, data analysis, quizzes, or other materials in this course are plagiarized or have been shared between students (current or past), no credit will be given for the work in question. Cases of suspect academic dishonesty will be handled according to University guidelines.

Grading Policy:

The University uses the +/- grading scale system and it is implemented in this course. Grades are weighted. Grade rounding only applies to the final course grade percentage. Sakai reports course grades to TWO digits past the decimal (XX.XX%); this percentage is rounded to the closest integer. For example, an 89.50% or 89.90% (B+) rounds up to a 90% (A-), BUT an 89.30% or 89.45% (B+) round to the integer 89% (B+), as it is the closest integer. There are no extra credit assignments in Chem 214 because, frankly, there is nothing of the sort in the "real world."

If you do not turn in work for 2 or more lab experiments (synchronous and asynchronous), you cannot pass the course [I have to report the situation to the Dean's Office, Wellness Center, and academic advisor the lack of coursework completion]. There is no final exam in this course.

Grading Scale* (%): *subject to change at Lab Coordinator discretion.

Course Grade %	Letter Grade
94 – 100	А
90 – 93	A-
87 – 89	B+
83 – 86	В
80 – 82	B-
77 – 79	C+
73 – 76	С
70 – 72	C-
60 – 69	D
0 – 59	F

Late Work Policies:

QUIZZES: If not completed on time, a 24-hour grace period is allotted to take the quiz past its due date [1pt penalty applied to late quiz submission]. After the 24-hour grace period, if a quiz is not taken a 0 is the final grade. Quizzes cannot be accessed after the late 24-hour grace period as answers are automatically programmed to be released then.

NOTEBOOK ENTRIES: If not completed on time, a 48-hour grace period is allotted to turn the work in late [2pt penalty for lateness applied to grade]. After 1-week, if the notebook is not turned in a 0 is the final grade.

IN-LAB DATA & ANALYSIS OR ONLINE LAB ACTIVITY RESULTS: If not completed on time, a 48-hour grace period is allotted to turn the work in late [5pt penalty for lateness applied to grade]. After 48-hours, if not turned in a 0 is the final grade.

FORMAL LAB REPORT, first draft for peer review: If not completed on time, a 24-hour grace period is allotted to turn the work in late. After 24-hours, if the lab report draft is not turned in a 0 / 20 is the final grade for the peer review portion of the report grade as a student cannot participate in the peer review if they themselves do not turn in a formal lab report draft.

FORMAL LAB REPORT, final draft (post-peer review): If not completed on time, a 72-hour grace period is allotted to turn the work in late [4pts penalty for every 12hrs past the due date/time]. After a full 72-hours past the due date/time, if the lab report draft is not turned in a 0 is the final grade for the final draft lab report grade.

SOCIAL JUSTICE FORUM/DISCUSSION POSTS: Late posting not possible. The forum is open almost all term. Once the forum/discussion closes it cannot be accessed.

See next two pages for itemized list of graded course work items.

Course Work Point Breakdown:

Course Activities	Origin or Sakai (Location of Submission)	Points	Weighted % of Final Grade
LAB EXPERIMENT / ONLINE LAB ACTIVITY			
WEEK 1: Acid-Base Titration Experiment: KHP & NaOH. %KHP Unknown (inperson lab experiment) Note: accuracy of data collected/lab results weighted in this grade.	Lab Results submitted to Sakai (Assignments)	100	
WEEK 1: Penny Statistics Using Excel (online homework; data analysis) Notes: student generate stats results when completing the activity. Accuracy of the work is taken into account for this grade.	Excel File submitted to Sakai (Assignments)	100	40%
WEEK 2: Polyprotic Acid Titration Experiment (in-person lab experiment) Note: accuracy of data collected/lab results weighted in this grade.	Lab Results submitted in Excel to Sakai (Assignments)	100	
WEEK 2: Buffers Lab (in person lab experiment) Notes: Students prepare buffers in several ways, adjust pH of buffers, etc.	Lab Results submitted to Sakai (Assignments)	100	

WEEK 3: EDTA Titration Analysis of Water Total Hardness; Ion Chromatography Analysis of Water Experiment (in-person lab experiment)	Lab Results submitted to Sakai (Assignments)	100	
Note: accuracy of data collected/lab results weighted in this grade. WEEK 4: HPLC Simulator (online lab activity; instrument parameters)	Excel or Word Doc or PDF submitted to		
Notes: Students use the Excel simulation, try various parameters to optimize separation. Document how parameter changes results.	Sakai (Assignments)	100	
WEEK 4: HPLC in Practice (in-person lab experiment) Note: accuracy of data collected/lab results weighted in this grade.	Lab Results submitted to Sakai (Assignments)	100	
WEEK 5: UV-Vis: External Standards Iron Analysis Experiment (in person lab experiment) Note: accuracy of data collected/lab results weighted in this grade.	Lab Results in Excel submitted to Sakai (Assignments)	100	
WEEK 5: UV-Vis: Ext. Std & Standard Addition, Food Dye (in-person lab experiment)	Lab Results in Excel File submitted to Sakai (Assignments)	100	
Note: accuracy of data collected/lab results weighted in this grade. WEEK 6: ATR-FTIR Quantitative Analysis (in-person lab experiment) Note: accuracy of data collected/lab results weighted in this grade.	Lab Results in Excel File submitted to Sakai (Assignments)	100	
QUIZZES (Tests and Quizzes)	Jakai (Assigninents)		
Quiz on Syllabus and Safety	Sakai (Tests & Quizzes)	20	
Quiz on Microsoft Excel and Stats	Sakai (Tests & Quizzes)	20	
Quiz on Buffers	Sakai (Tests & Quizzes)	20	15%
Quiz on Titrations	Sakai (Tests & Quizzes)	20	1370
Quiz on Chromatography Methods	Sakai (Tests & Quizzes)	20	
Quiz on UV-Vis Iron, Calibration Curves, Standard Solutions	Sakai (Tests & Quizzes)	20	
NOTEBOOK ENTRIES (use CamScanner, submit each notebook en			
All in-person experiments and activities need a notebook entry; these labs			
Notebook Entry for Acid-Base Titration (KHP and NaOH) Lab	Sakai (Assignments)	30	
Notebook Entry for Buffers Lab	Sakai (Assignments)	30	15%
Notebook Entry for EDTA Titration Water Analysis Lab	Sakai (Assignments)	30	1370
Notebook Entry for UV-Vis: Ext. Std., Iron Analysis Lab	Sakai (Assignments)	30	
SOCIAL JUSTICE			
Discussion/Forums Posting identifying Social Justice issues in the sciences (Sakai activity)	Sakai (Discussion)	30	5%
LAB REPORT (typed and submitted as Word or PDF file)			
WEEK 3: Formal Lab Report Acid-Base Titration [PDF or Word Doc] rough draft for the peer review / participation in the peer review	Sakai (Assignments)	20	15%
WEEK 4: Formal Lab Report Acid-Base Titration [PDF or Word Doc] final draft	Sakai (Assignments)	200	
LAB SAFETY	1		
Safety and lab clean-up points for in-person sessions. Deductions for unsafe action or lateness apply to this grade.	In Lab	20	10%
			100%

Lab Report and Notebook Grading Rubrics:

Lab Report	Points

Title Page	20
Introduction/Purpose	45
Results and Data	50
Discussion	30
Conclusion	20
Grammar/Formatting/Spelling	25
Proper File Type (Word or PDF) and	10
Location Submission (Sakai)	10
Total	200

Notebook Entries (Each notebook entry is graded with this rubric)	Points
Table of Contents (experiment/activity title & page numbers listed). This is not graded	N/A
but for the experimenter's benefit to keep track of notebook entries.	
Student Name, Section #, Date at the top of each notebook page. Notebook is not	N/A
graded if this information is not on all scanned notebook pages for each entry.	
Title of Experiment/Activity clearly defined on first page of notebook entry	2
Introduction Section	5
Results and Data Section [Raw Data and Calculations, Graphs, Tables, Etc.] Each lab	13
experiment and/or online lab activity has different results and data processing.	
Conclusion Section	7
Organization and Proofreading (sections clearly labeled, writing legible, sentences	3
complete and spelling/grammar ok, etc.)	
Total	30

Additional Student Resources:

A considerable amount of technology is utilized in this course. Here are links of information guides in the event that students need more structured guidance on using the tools in the course in order to be successful. A link to the University Help Desk is also provided for technology questions. Students can email the Lab Coordinator and TAs about various University information, but the links below may reveal the answer more quickly when a student reads them on their own. These guides are written by the pros. Use links any time additional University info is needed.

Career Services

Coronavirus Updates from University

First and Second Year Advising

<u>Information Technology Service Desk</u> (ITS Help Desk)

Panopto Information

Resource Guide for Online Learning

SAKAI student guide

Success Coaching

Student Accessibility Center

Tutoring Center

Writing Center

ZOOM Information and Contacting ZOOM Support

DISCLAIMER: Dr. Binaku reserves the right to revise this syllabus to correct any unintentional mistakes found at any point of the term. Students will be notified if any changes have been made.

COPYRIGHT DISCLAIMER: All portions of the Chem 214 syllabus, course materials in Sakai (PowerPoints, all handouts (Word or PDF), rubrics, directions, lab videos) are NOT allowed for distribution outside of class nor outside of the University. Uploading, posting, copying, or sharing electronic/non-electronic Chem 214 materials outside of class [i.e. share sites] is NOT allowed. If it is discovered a student completes such action, the University will be notified immediately as that is breaking copyright law.

See the remaining pages of the syllabus for lab schedule and homework due date(s) information!

CALENDAR MAP of the lab course.

Month	Mon	Tue	Wed	Thu	Fri
May 2023	22	23 First Day Intro, Safety, Glassware/Equipment Use, Calc. Review Lab: Solutions Preparation	24	25 Lab: Acid Base Titration (KHP and NaOH) Homework Lab: Penny Stats	26
	29	30 Lab: Polyprotic Acid Titration	31	1 Lab: Buffers	2
June 2023	5	6 Lab: EDTA Titration Hard Water and Ion Chromatography Analysis of Water	7	8 Formal lab report draft due today Lab: EDTA Titration Hard Water and Ion Chromatography Analysis of Water	9

Month	Mon	Tue	Wed	Thu	Fri
	12	Asynchronous today: Read Peer Review & Revise Lab Report AND Asynchronous HPLC Lab Simulation	14	15 Formal lab report final draft due today Lab: HPLC in Practice	16
	19	20 Lab: UV-Vis Iron in Water	21	22 Lab: UV-Vis % Tartrazine	23
	26	27 Lab: ATR-FTIR Quantification of Eugenol AND	28	29 Last Day Check Out, Careers in Chemistry And	30
		Solution Cleanup		Make Up Lab Opportunity	

See the remaining pages of the syllabus for lab schedule and homework due date(s) information!

WEEK &	Lecture, Lab Activity, & Homework*
Dates	*activity opens on the class day it is listed
	Lecture to Watch BEFORE Class Meets:
	Intro & Syllabus Lecture (Panopto)
	4 videos (Online Lab Activities): Lab Techniques, Common Glassware and Uses, Solutions &
	Concentrations, and Making Solutions in Lab
	Lab Manual Reading BEFORE Class Meets: Read Lab Manual Acid-Base Titration Lab for NaOH
WEEK 1	soln prep AND read EDTA Lab for EDTA soln prep
	Lecture Content in Class:
	Glassware, Equipment, & Safety Lecture, Calc Review, Solution Prep
Tuesday	Lab Activity/Experiment in Class:
Tuesday, May 23	Equipment check-in, calculation review, Q & A, solution preparation (NaOH soln and EDTA soln)
	Homework in Sakai (listed below), due by next lab period, Thursday, May 25 by 8:30am:
	1. Read syllabus if you haven't yet AND explore the features of the Sakai course site. Purchase
	lab coat, notebook, calculator.
	2. Practice submitting notebook pages, submit review sheet answers as PDF file to Sakai (Assignments)

	 Watch Intro to Lab Notebook Writing Lecture (<i>Panopto</i>), read related PDF documents (<i>Resources</i>) Quiz on Syllabus, Course Req., Safety (<i>Tests & Quizzes</i>) Watch Panopto: Acid-Base Titration: KHP and NaOH. Determine %KHP in Unknown (<i>Panopto</i>) Read in Lab Manual Acid-Base Titration: KHP and NaOH. Determine %KHP in Unknown Write Date, Title, Introduction sections for acid-base titration lab in notebook. Must be done
	before walking into lab on Thurs.! See next page for Thursday's agenda
WEEK &	
	Lecture, Lab Activity, & Homework* *activity opens on the class day it is listed
Dates	Lecture, Lab Activity, & Homework* *activity opens on the class day it is listed Lecture Content in Class: Acid-Base Titration (KHP and NaOH) Tips
	*activity opens on the class day it is listed Lecture Content in Class:
Dates	*activity opens on the class day it is listed Lecture Content in Class: Acid-Base Titration (KHP and NaOH) Tips Lab Activity/Experiment in Class:
	*activity opens on the class day it is listed Lecture Content in Class: Acid-Base Titration (KHP and NaOH) Tips Lab Activity/Experiment in Class: Complete Lab Experiment Acid-Base Titration: KHP and NaOH. Determine %KHP in Unknown

	Lecture Content in Class:
	Polyprotic Acid Titration Tips
	Lab Activity/Experiment in Class:
	Complete Lab Experiment Polyprotic Acid Titration
	Homework in Sakai (listed below), due by next lab period Thursday, June 1 by 8:30am:
WEEK 2 Tuesday, May 30	 Complete calculations for lab data and generate the graphs (titration curve, 1st derivative, 2nd derivative) in Excel. Submit the required results to Sakai (Assignments) Watch 4 videos on buffers (Online Lab Activities) Watch Panopto: Buffers (Panopto) Read in Lab Manual the Buffer Lab Write Date, Title, Introduction sections for Buffer lab in lab notebook. Must be done before walking into lab on Thurs.!
WEEK & Dates	Lecture, Lab Activity, & Homework* *activity opens on the class day it is listed
	Lecture Content in Class:
	Buffer Preparation Tips
	Lab Astinity/Franching out in Class.
	Lab Activity/Experiment in Class:
WEEK 2	
	Complete Lab Experiment on Buffers
	Homework in Sakai (listed below), due by next lab period Tuesday, June 6 by 8:30am:
Thursday, June 1	 Homework in Sakai (listed below), due by next lab period Tuesday, June 6 by 8:30am: Submit Buffer lab results to Sakai (Assignments) Submit completed Notebook Entry for Buffers to Sakai (Assignments) Watch Panopto: EDTA Titration and Ion Chromatography (Panopto) Watch 5 content related videos on IC posted in Sakai (Online Lab Activities) Read in Lab Manual the EDTA Titration & Water Hardness via IC Write Date, Title, Introduction sections for EDTA titration and IC in lab notebook. Must be done before walking in lab on Tues.! Obtain your own water sample for lab's analysis!
June 1	 Homework in Sakai (listed below), due by next lab period Tuesday, June 6 by 8:30am: Submit Buffer lab results to Sakai (Assignments) Submit completed Notebook Entry for Buffers to Sakai (Assignments) Watch Panopto: EDTA Titration and Ion Chromatography (Panopto) Watch 5 content related videos on IC posted in Sakai (Online Lab Activities) Read in Lab Manual the EDTA Titration & Water Hardness via IC Write Date, Title, Introduction sections for EDTA titration and IC in lab notebook. Must be done before walking in lab on Tues.!
	 Homework in Sakai (listed below), due by next lab period Tuesday, June 6 by 8:30am: Submit Buffer lab results to Sakai (Assignments) Submit completed Notebook Entry for Buffers to Sakai (Assignments) Watch Panopto: EDTA Titration and Ion Chromatography (Panopto) Watch 5 content related videos on IC posted in Sakai (Online Lab Activities) Read in Lab Manual the EDTA Titration & Water Hardness via IC Write Date, Title, Introduction sections for EDTA titration and IC in lab notebook. Must be done before walking in lab on Tues.! Obtain your own water sample for lab's analysis!
June 1	Homework in Sakai (listed below), due by next lab period Tuesday, June 6 by 8:30am: 1. Submit Buffer lab results to Sakai (Assignments) 2. Submit completed Notebook Entry for Buffers to Sakai (Assignments) 3. Watch Panopto: EDTA Titration and Ion Chromatography (Panopto) 4. Watch 5 content related videos on IC posted in Sakai (Online Lab Activities) 5. Read in Lab Manual the EDTA Titration & Water Hardness via IC 6. Write Date, Title, Introduction sections for EDTA titration and IC in lab notebook. Must be done before walking in lab on Tues.! 7. Obtain your own water sample for lab's analysis! Lecture Content in Class: EDTA Titration Total Hardness, IC Analysis Tips
June 1	Homework in Sakai (listed below), due by next lab period Tuesday, June 6 by 8:30am: 1. Submit Buffer lab results to Sakai (Assignments) 2. Submit completed Notebook Entry for Buffers to Sakai (Assignments) 3. Watch Panopto: EDTA Titration and Ion Chromatography (Panopto) 4. Watch 5 content related videos on IC posted in Sakai (Online Lab Activities) 5. Read in Lab Manual the EDTA Titration & Water Hardness via IC 6. Write Date, Title, Introduction sections for EDTA titration and IC in lab notebook. Must be done before walking in lab on Tues.! 7. Obtain your own water sample for lab's analysis! Lecture Content in Class: EDTA Titration Total Hardness, IC Analysis Tips Lab Activity/Experiment in Class:
June 1 WEEK 3	Homework in Sakai (listed below), due by next lab period Tuesday, June 6 by 8:30am: 1. Submit Buffer lab results to Sakai (Assignments) 2. Submit completed Notebook Entry for Buffers to Sakai (Assignments) 3. Watch Panopto: EDTA Titration and Ion Chromatography (Panopto) 4. Watch 5 content related videos on IC posted in Sakai (Online Lab Activities) 5. Read in Lab Manual the EDTA Titration & Water Hardness via IC 6. Write Date, Title, Introduction sections for EDTA titration and IC in lab notebook. Must be done before walking in lab on Tues.! 7. Obtain your own water sample for lab's analysis! Lecture Content in Class: EDTA Titration Total Hardness, IC Analysis Tips Lab Activity/Experiment in Class: Complete Lab Experiment EDTA Titration of Water Sample and/or Ion Chromatography Analysis
June 1	Homework in Sakai (listed below), due by next lab period Tuesday, June 6 by 8:30am: 1. Submit Buffer lab results to Sakai (Assignments) 2. Submit completed Notebook Entry for Buffers to Sakai (Assignments) 3. Watch Panopto: EDTA Titration and Ion Chromatography (Panopto) 4. Watch 5 content related videos on IC posted in Sakai (Online Lab Activities) 5. Read in Lab Manual the EDTA Titration & Water Hardness via IC 6. Write Date, Title, Introduction sections for EDTA titration and IC in lab notebook. Must be done before walking in lab on Tues.! 7. Obtain your own water sample for lab's analysis! Lecture Content in Class: EDTA Titration Total Hardness, IC Analysis Tips Lab Activity/Experiment in Class:

1. Work on calculations for the parts of the experiment that were completed. Whatever you didn't finish wet chem wise, must be finished by end of lab on Thursday. 2. Watch Panopto lecture: Formal Lab Reports (Panopto), review PDF handout (Resources) 3. Type a formal lab report draft on the Acid-Base Titration: KHP & NaOH %KHP in Unknown lab experiment and submit as a PDF or Word file to Sakai (Assignments) Lecture Content in Class: None. All lab work today! Lab Activity/Experiment in Class: FINISH Lab Experiment EDTA Titration and/or lon Chromatography Analysis of Water Sample Homework in Sakai (listed below), due by next lab period Tuesday, June 13 by 8:30am: 1. Complete calculations for EDTA titration total hardness. Complete calculations for IC total hardness. Submit the required results to Sakai (Assignments) 3. Take the quiz on Buffers in Sakai (Tests and Quizzes) 3. Take the quiz on Buffers in Sakai (Tests and Quizzes) 5. Complete the peer review assigned to you for the Acid-Base Titration: KHP & NaOH %KHP in Unknown formal report in Sakai (Assignments) 6. Check over graded work thus far: Contact Dr. Binaku with any grading questions or if you see a discrepancy/error. We are halfway through summer session course! 7. Make 1 original post in social justice forum in Sakai (Discussions) if you haven't yet. Lecture, Lab Activity, & Homework* **activity opens on the class day it is listed Asynchronous Day –aka you don't need to come to lab Complete the HPLC Simulator Homework in Sakai (listed below), due by next lab period Thursday, June 15 by 8:30am: 1. Quiz on Titrations in Sakai (Tests & Quizzes) 2. Wartch Panopto: HPLC Simulation (Panopto) and read Lab Manual info 3. Download HPLC simulator activity, Follow manual directions. Submit Word Doc with info of all results to Sakai (Assignments) 5. Review peer review comments, make revisions to formal lab report draft, and submit FINAL DRAF1 of Acid-Base Titration. KHP & NaOH %KHP in Unknown lab experiment and submit as a PDF or Word file to Sa		
Homework in Sakai (listed below), due by next lab period Tuesday, June 13 by 8:30am: 1. Complete calculations for EDTA titration total hardness. Complete calculations for IC total hardness. Submit the required results to Sakai (Assignments) 2. Submit Completed Notebook Entry for EDTA Titration and IC Analysis to Sakai (Assignments) 3. Take the quiz on Buffers in Sakai (Tests and Quizzes) 4. Watch Peer Review (Panopto) and/or read the PDF regarding the Peer Review process. 5. Complete the peer review assigned to you for the Acid-Base Titration: KHP & NaOH %KHP in Unknown formal report in Sakai (Assignments) 6. Check over graded work thus far. Contact Dr. Binaku with any grading questions or if you see a discrepancy/error. We are halfway through summer session course! 7. Make 1 original post in social justice forum in Sakai (Discussions) if you haven't yet. WEEK & Dates Asynchronous Day -aka you don't need to come to lab Complete the HPLC Simulator Homework in Sakai (listed below), due by next lab period Thursday, June 15 by 8:30am: 1. Quiz on Titrations in Sakai (Tests & Quizzes) 2. Watch Panopto: HPLC Simulation (Panopto) and read Lab Manual info 3. Download HPLC simulator Excel file in Sakai (Online Lab Activities) 4. Complete HPLC simulator Excel file in Sakai (Online Lab Activities) 5. Review peer review comments, make revisions to formal lab report draft, and submit FINAL DRAFT of Acid-Base Titration: KHP & NaOH %KHP in Unknown lab experiment and submit as a PDF or Word file to Sakai (Assignments) Lecture Content in Class:		didn't finish wet chem wise, must be finished by end of lab on Thursday. 2. Watch Panopto lecture: Formal Lab Reports (Panopto), review PDF handout (Resources) 3. Type a formal lab report draft on the Acid-Base Titration: KHP & NaOH %KHP in Unknown lab experiment and submit as a PDF or Word file to Sakai (Assignments) Lecture Content in Class: None. All lab work today! Lab Activity/Experiment in Class:
Thursday, June 8 Thursday, June 9 Thursday, June 9 Thursday, June 19 Thursday, June 19 Thursday, June 19 Thursday, June 19 Thursday, June 15	WEEK 3	
WEEK & Dates Complete the HPLC Simulator		hardness. Submit the required results to Sakai (Assignments) 2. Submit Completed Notebook Entry for EDTA Titration and IC Analysis to Sakai (Assignments) 3. Take the quiz on Buffers in Sakai (Tests and Quizzes) 4. Watch Peer Review (Panopto) and/or read the PDF regarding the Peer Review process. 5. Complete the peer review assigned to you for the Acid-Base Titration: KHP & NaOH %KHP in Unknown formal report in Sakai (Assignments) 6. Check over graded work thus far. Contact Dr. Binaku with any grading questions or if you see a discrepancy/error. We are halfway through summer session course!
#activity opens on the class day it is listed Asynchronous Day –aka you don't need to come to lab Complete the HPLC Simulator Homework in Sakai (listed below), due by next lab period Thursday, June 15 by 8:30am: 1. Quiz on Titrations in Sakai (Tests & Quizzes) 2. Watch Panopto: HPLC Simulation (Panopto) and read Lab Manual info 3. Download HPLC simulator excel file in Sakai (Online Lab Activities) 4. Complete HPLC simulator activity. Follow manual directions. Submit Word Doc with info of all results to Sakai (Assignments) 5. Review peer review comments, make revisions to formal lab report draft, and submit FINAL DRAFT of Acid-Base Titration: KHP & NaOH %KHP in Unknown lab experiment and submit as a PDF or Word file to Sakai (Assignments) Lecture Content in Class: HPLC in Practice WEEK 4 Complete Lab Experiment Using HPLC Homework in Sakai (listed below), due by next lab period Tuesday, June 20 by 8:30am: 1. Submit the required results to Sakai (Assignments) 2. Watch Panopto: UV-VIS Iron Analysis (Panopto)	WEEK 8	
WEEK 4 Asynchronous Day —aka you don't need to come to lab Complete the HPLC Simulator Homework in Sakai (listed below), due by next lab period Thursday, June 15 by 8:30am: 1. Quiz on Titrations in Sakai (Tests & Quizzes) 2. Watch Panopto: HPLC Simulation (Panopto) and read Lab Manual info 3. Download HPLC simulator Excel file in Sakai (Online Lab Activities) 4. Complete HPLC simulator activity. Follow manual directions. Submit Word Doc with info of all results to Sakai (Assignments) 5. Review peer review comments, make revisions to formal lab report draft, and submit FINAL DRAFT of Acid-Base Titration: KHP & NaOH %KHP in Unknown lab experiment and submit as a PDF or Word file to Sakai (Assignments) Lecture Content in Class: HPLC in Practice WEEK 4 Complete Lab Experiment Using HPLC Homework in Sakai (listed below), due by next lab period Tuesday, June 20 by 8:30am: 1. Submit the required results to Sakai (Assignments) 2. Watch Panopto: UV-VIS Iron Analysis (Panopto)		Lecture, Lab Activity, & Homework
Tuesday, June 13 1. Quiz on Titrations in Sakai (Tests & Quizzes) 2. Watch Panopto: HPLC Simulation (Panopto) and read Lab Manual info 3. Download HPLC simulator Excel file in Sakai (Online Lab Activities) 4. Complete HPLC simulator activity. Follow manual directions. Submit Word Doc with info of all results to Sakai (Assignments) 5. Review peer review comments, make revisions to formal lab report draft, and submit FINAL DRAFT of Acid-Base Titration: KHP & NaOH %KHP in Unknown lab experiment and submit as a PDF or Word file to Sakai (Assignments) Lecture Content in Class: HPLC in Practice WEEK 4 Thursday, June 15 1. Submit the required results to Sakai (Assignments) 2. Watch Panopto: UV-VIS Iron Analysis (Panopto)	Dates	*activity opens on the class day it is listed
Tuesday, June 13 1. Quiz on Titrations in Sakai (Tests & Quizzes) 2. Watch Panopto: HPLC Simulation (Panopto) and read Lab Manual info 3. Download HPLC simulator Excel file in Sakai (Online Lab Activities) 4. Complete HPLC simulator activity. Follow manual directions. Submit Word Doc with info of all results to Sakai (Assignments) 5. Review peer review comments, make revisions to formal lab report draft, and submit FINAL DRAFT of Acid-Base Titration: KHP & NaOH %KHP in Unknown lab experiment and submit as a PDF or Word file to Sakai (Assignments) Lecture Content in Class: HPLC in Practice WEEK 4 Lab Activity/Experiment in Class: Complete Lab Experiment Using HPLC Thursday, June 15 1. Submit the required results to Sakai (Assignments) 2. Watch Panopto: UV-VIS Iron Analysis (Panopto)	Dates	*activity opens on the class day it is listed
Tuesday, June 13 1. Quiz on Titrations in Sakai (Tests & Quizzes) 2. Watch Panopto: HPLC Simulation (Panopto) and read Lab Manual info 3. Download HPLC simulator Excel file in Sakai (Online Lab Activities) 4. Complete HPLC simulator activity. Follow manual directions. Submit Word Doc with info of all results to Sakai (Assignments) 5. Review peer review comments, make revisions to formal lab report draft, and submit FINAL DRAFT of Acid-Base Titration: KHP & NaOH %KHP in Unknown lab experiment and submit as a PDF or Word file to Sakai (Assignments) Lecture Content in Class: HPLC in Practice WEEK 4 Lab Activity/Experiment in Class: Complete Lab Experiment Using HPLC Thursday, June 15 1. Submit the required results to Sakai (Assignments) 2. Watch Panopto: UV-VIS Iron Analysis (Panopto)		Asynchronous Day –aka you don't need to come to lab
WEEK 4 Lab Activity/Experiment in Class: Complete Lab Experiment Using HPLC Thursday, June 15 1. Submit the required results to Sakai (Assignments) 2. Watch Panopto: UV-VIS Iron Analysis (Panopto)		Asynchronous Day —aka you don't need to come to lab Complete the HPLC Simulator
WEEK 4 Lab Activity/Experiment in Class: Complete Lab Experiment Using HPLC Thursday, June 15 1. Submit the required results to Sakai (Assignments) 2. Watch Panopto: UV-VIS Iron Analysis (Panopto)	WEEK 4 Tuesday,	Asynchronous Day –aka you don't need to come to lab Complete the HPLC Simulator Homework in Sakai (listed below), due by next lab period Thursday, June 15 by 8:30am: 1. Quiz on Titrations in Sakai (Tests & Quizzes) 2. Watch Panopto: HPLC Simulation (Panopto) and read Lab Manual info 3. Download HPLC simulator Excel file in Sakai (Online Lab Activities) 4. Complete HPLC simulator activity. Follow manual directions. Submit Word Doc with info of all results to Sakai (Assignments) 5. Review peer review comments, make revisions to formal lab report draft, and submit FINAL DRAFT of Acid-Base Titration: KHP & NaOH %KHP in Unknown lab experiment and submit as a
Lab Activity/Experiment in Class: Complete Lab Experiment Using HPLC Thursday, June 15 1. Submit the required results to Sakai (Assignments) 2. Watch Panopto: UV-VIS Iron Analysis (Panopto)	WEEK 4 Tuesday,	Asynchronous Day —aka you don't need to come to lab Complete the HPLC Simulator Homework in Sakai (listed below), due by next lab period Thursday, June 15 by 8:30am: 1. Quiz on Titrations in Sakai (Tests & Quizzes) 2. Watch Panopto: HPLC Simulation (Panopto) and read Lab Manual info 3. Download HPLC simulator Excel file in Sakai (Online Lab Activities) 4. Complete HPLC simulator activity. Follow manual directions. Submit Word Doc with info of all results to Sakai (Assignments) 5. Review peer review comments, make revisions to formal lab report draft, and submit FINAL DRAFT of Acid-Base Titration: KHP & NaOH %KHP in Unknown lab experiment and submit as a PDF or Word file to Sakai (Assignments)
Complete Lab Experiment Using HPLC Thursday, June 15 1. Submit the required results to Sakai (Assignments) 2. Watch Panopto: UV-VIS Iron Analysis (Panopto)	WEEK 4 Tuesday,	Asynchronous Day —aka you don't need to come to lab Complete the HPLC Simulator Homework in Sakai (listed below), due by next lab period Thursday, June 15 by 8:30am: 1. Quiz on Titrations in Sakai (Tests & Quizzes) 2. Watch Panopto: HPLC Simulation (Panopto) and read Lab Manual info 3. Download HPLC simulator Excel file in Sakai (Online Lab Activities) 4. Complete HPLC simulator activity. Follow manual directions. Submit Word Doc with info of all results to Sakai (Assignments) 5. Review peer review comments, make revisions to formal lab report draft, and submit FINAL DRAFT of Acid-Base Titration: KHP & NaOH %KHP in Unknown lab experiment and submit as a PDF or Word file to Sakai (Assignments) Lecture Content in Class:
Complete Lab Experiment Using HPLC Thursday, June 15 1. Submit the required results to Sakai (Assignments) 2. Watch Panopto: UV-VIS Iron Analysis (Panopto)	WEEK 4 Tuesday,	Asynchronous Day —aka you don't need to come to lab Complete the HPLC Simulator Homework in Sakai (listed below), due by next lab period Thursday, June 15 by 8:30am: 1. Quiz on Titrations in Sakai (Tests & Quizzes) 2. Watch Panopto: HPLC Simulation (Panopto) and read Lab Manual info 3. Download HPLC simulator Excel file in Sakai (Online Lab Activities) 4. Complete HPLC simulator activity. Follow manual directions. Submit Word Doc with info of all results to Sakai (Assignments) 5. Review peer review comments, make revisions to formal lab report draft, and submit FINAL DRAFT of Acid-Base Titration: KHP & NaOH %KHP in Unknown lab experiment and submit as a PDF or Word file to Sakai (Assignments) Lecture Content in Class:
Thursday, June 15 Homework in Sakai (listed below), due by next lab period Tuesday, June 20 by 8:30am: 1. Submit the required results to Sakai (Assignments) 2. Watch Panopto: UV-VIS Iron Analysis (Panopto)	WEEK 4 Tuesday, June 13	Asynchronous Day —aka you don't need to come to lab Complete the HPLC Simulator Homework in Sakai (listed below), due by next lab period Thursday, June 15 by 8:30am: 1. Quiz on Titrations in Sakai (Tests & Quizzes) 2. Watch Panopto: HPLC Simulation (Panopto) and read Lab Manual info 3. Download HPLC simulator Excel file in Sakai (Online Lab Activities) 4. Complete HPLC simulator activity. Follow manual directions. Submit Word Doc with info of all results to Sakai (Assignments) 5. Review peer review comments, make revisions to formal lab report draft, and submit FINAL DRAFT of Acid-Base Titration: KHP & NaOH %KHP in Unknown lab experiment and submit as a PDF or Word file to Sakai (Assignments) Lecture Content in Class: HPLC in Practice
3. Read in Lab Manual the UV-VIS Iron Analysis	WEEK 4 Tuesday, June 13	Asynchronous Day —aka you don't need to come to lab Complete the HPLC Simulator Homework in Sakai (listed below), due by next lab period Thursday, June 15 by 8:30am: 1. Quiz on Titrations in Sakai (Tests & Quizzes) 2. Watch Panopto: HPLC Simulation (Panopto) and read Lab Manual info 3. Download HPLC simulator Excel file in Sakai (Online Lab Activities) 4. Complete HPLC simulator activity. Follow manual directions. Submit Word Doc with info of all results to Sakai (Assignments) 5. Review peer review comments, make revisions to formal lab report draft, and submit FINAL DRAFT of Acid-Base Titration: KHP & NaOH %KHP in Unknown lab experiment and submit as a PDF or Word file to Sakai (Assignments) Lecture Content in Class: HPLC in Practice Lab Activity/Experiment in Class:
	WEEK 4 Tuesday, June 13 WEEK 4 Thursday,	Asynchronous Day —aka you don't need to come to lab Complete the HPLC Simulator Homework in Sakai (listed below), due by next lab period Thursday, June 15 by 8:30am: 1. Quiz on Titrations in Sakai (Tests & Quizzes) 2. Watch Panopto: HPLC Simulation (Panopto) and read Lab Manual info 3. Download HPLC simulator Excel file in Sakai (Online Lab Activities) 4. Complete HPLC simulator activity. Follow manual directions. Submit Word Doc with info of all results to Sakai (Assignments) 5. Review peer review comments, make revisions to formal lab report draft, and submit FINAL DRAFT of Acid-Base Titration: KHP & NaOH %KHP in Unknown lab experiment and submit as a PDF or Word file to Sakai (Assignments) Lecture Content in Class:

	 4. Write Date, Title, Introduction sections for UV-Vis iron lab in notebook. Must be done before walking into lab on Tues.! 5. Watch 4 videos (Online Lab Activities): UV-Vis Spec, Beer's Law, Intro to Spectrophotometer, and Calibration Curves
	Lecture Content in Class:
	UV-Vis Iron Analysis, External Standards Tips
	Lab Activity/Experiment in Class:
	Complete Lab Experiment UV-Vis Iron Analysis via External Standards
WEEK 5	Homework in Sakai (listed below), due by next lab period Thursday, June 22 by 8:30am:
	1. Complete UV-Vis Iron lab calculations and calibration curve in Excel. Submit Excel file results to Sakai (Assignments)
Tuesday,	2. Submit Completed Notebook Entry for UV-Vis iron lab to Sakai (Assignments)
June 20	3. Quiz on Chromatography in Sakai (<i>Tests & Quizzes</i>)
Julie 20	
	4. Watch Panopto: UV-VIS Artificial Dye (Panopto)
	5. Read in Lab Manual the UV-VIS Artificial Dye info
	6. Write Date, Title, Introduction sections for UV-Vis Artificial Dye in lab notebook. Must be
	done before walking into lab on Thurs.!
WEEK &	Lecture, Lab Activity, & Homework*
Dates	*activity opens on the class day it is listed
	Lecture Content in Class:
	UV-Vis % Tartrazine, External Standards and Standard Addition Lab Tips
	Lab Activity/Experiment in Class:
	Complete Lab Experiment UV-Vis % Tartrazine Analysis via External Standards and Standard
WEEK 5	Addition
	Homework in Sakai (listed below), due by next lab period Tuesday, June 27 by 8:30am:
Thursday,	1. Complete UV-Vis Tartrazine lab calculations and both calibration curves in Excel. Submit Excel
June 22	file results to Sakai (Assignments)
	2. Quiz on UV-Vis, Calibration Curves, Standard Solutions in Sakai (Tests & Quizzes)
	3. Watch Panopto: ATR-FTIR Quantitative Analysis (Panopto)
	4. Read in Lab Manual the ATR-FTIR Analysis info
	5. Write date, title, Introduction sections for ATR-FTIR Quantification in lab notebook. Must be
	done before walking into lab on Tues.!
	Lecture Content in Class:
	ATR-FTIR Quantification of Eugenol Tips
WEEK 6	Chemical Clean-Up
	Lab Activity/Experiment in Class:
Tuesday,	Complete Lab Experiment ATR-FTIR Quantification of Eugenol
June 27	Harris and the first of the fir
	Homework in Sakai (listed below), due by next lab period Thursday, June 29 by 8:30am:

	1. Complete ATR-FTIR lab calculations and both calibration curves in Excel. Submit Excel file results to Sakai (Assignments) before you leave lab today
	2. Make 2nd original post in social justice forum in Sakai (<i>Discussions</i>) OR both posts if you haven't done any yet.
	3. Check all grades in Sakai (Gradebook). Contact Dr. Binaku via email with any questions or if you see any discrepancies. Bring grading questions to class on Thursday too!
	LAST DAY OF CLASS
	Lab Equipment Checkout
	Research/Internship/Grad School/Careers Information
WEEK 6	*If new AA instrument arrives in time, we may do a new lab experiment today. TBA.
Thursday, June 29	MAKE UP LAB DAY: Only opportunity to make up a missed lab if absence occurred this term. You are done with class! BRAVO
	All grading questions must be resolved during class time, in-person. Grades are final and go in LOCUS today after class. Emails after class today about grades will be ignored.